

Model Studies on the Regeneration of Silica-Alumina Cracking Catalysts. I. Method of Study and the Coke Distribution in a Particle of Silica-Alumina Bead Catalyst, by D. P. Dobychin, Ia. M. Klibanova.

RUSSIAN, Izv. Akad. Nauk SSSR, Ser. Khim., Vol. XXXIII, No. 4, 1959, pp 869-876,

ATS RJ-1893

Sci - Chem
Dec 59

104,072

The Effect of the Conditions of Preparing Silica
and Alumina Hydrogels on the Structure and
Activity of Mixed Silica-Alumina Catalysts, by
M. A. Kaliko, et al.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXIII, No 4,
1959, pp 922-929,

ATS RJ-1894

Sci - Chem
Dec 59

104,073

The Study of the Mechanism of Electrochemical
Corrosion of Titanium, by N. D. Tomashov, R. M.
At'sevskiy.

RUSSIAN, per, Zhur. Fiz. Khim., Vol XXXIII, No 5,
1956, pp 781-783.

DSIR LLU RTS 1960

Sci .. Chem

Aug 60

122,509

Study on the Kinetics of Ion Exchange
Sorption II. The Kinetics of Exchange
With the Participation of Hydrogen Ions,
by O. P. Fedoseyeva, et al.

RUSSIAN, Izv. Akad. Nauk Fiz. Khim., Vol. XXXIII,
No. 4, 1959, pp. 936-942.

CSIRO

Sci - Phys
Jun 62

203,353

62-10229

Fedosseeva, O. P., Chernyaeva, E. P., and
Tunitskil, N. N.

STUDY OF THE KINETICS OF ION EXCHANGE
SORPTION. II. KINETICS OF EXCHANGE WITH
PARTICIPATION OF HYDROGEN IONS [Issledovanie
Kinetiki Ionobmennoi Sorbstii. II. Kinetika Obmen
s Uchashtiem Vodородnykh Ionov]. [1961] [9]p. 7 refs.
Order from OTS or SLA \$1.10 62-10229

Trans. of Zhurnal Fizicheskoi Khimii (USSR) 1959,
v. 33 (no. 4) p. 936-942.

Another trans. is available from LC or SLA m\$1.80,
ph\$1.80 as 61-13398, DSIR NLL M.2037 [1960] 8p.

DESCRIPTORS: *Ion exchange resins, Resins,
*Hydrogen, Sodium, Rubidium, *Ions, Exchange
reactions, Reaction kinetics, Adsorption, Diffusion.

For abstract see Technical Translations 5: 297, 1961.
(Chemistry--Physical, TT, v. 7, no. 10)

I. Fedosseeva, O. P.
II. Chernyaeva, E. P.
III. Tunitskil, N. N.
IV. Title: Kinetics. . .

Office of Technical Services

The Catalytic Properties of the System Pt + Au,
by A. A. Alchudzhan, M. A. Indzhikyan,

RUSSIAN, per. Zhur Fiz Khim, Vol XXXIII, No 5,
1959, pp 983-987.

ATS RJ-2159

Sci. - Chem

May 60

115,440

Aykazyan, E. A.
PASSIVATION OF THE PLATINUM ELECTRODE (O
Passivatsii Platinyogo Elektroda). Apr 61 [13]p.
16 refs. RTS 1797.

Order from OTS or SLA \$1.60

61-19671

Trans. of Zhurnal Fizicheskoy Khimii (USSR) 1959,
v. 33, no. 3, p. 1015-1022.

DESCRIPTORS: Electrodes, *Platinum electrodes,
Electrochemistry, *Passivity, *Polarization.

(Chemistry-Physical, TT, v. 6, no. 2)

61-19671

I. Aykazyan, E. A.
II. RTS-1797
III. Department of Scientific
and Industrial Research
(Gr. Brit.)

101502

Office of Technical Services

Model Studies of Regeneration of Silica-Alumina Cracking Catalysts. II. Burning-off of Surface Coke and General Kinetic Relationships of the Process, by D. P. Gobyshin, Ts. M. Klivanova, 8 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXIII, 1959, pp 1023-1029.

Sci
Vol. IV, No 2
Mar 62

ATS-06137R
ATS-RJ-2380
188, 885

Acidity Function of the Methyl Ether-BF₃ System,
by G. B. Manelis.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXIII, No 5,
1959, pp 1030-1034.

NLL RRB 2232

Sci -
Jul 63

238,182

Self-Diffusion of Dibasic Phosphate Ion in the
System Water-Electrolyte-Nonelectrolyte, by
P. Ya. Siver, 9 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXIII, No 5, 1959,
pp 1065-1070.

SLA 60-17727

Sci

129,408

OTS, Vol. IV, No 3

Sep 60

The Photocatalytic Properties of Zinc Oxide in
the Reduction of Methylene Blue. 4. Poisoning
of the Photocatalysts by Nitro Compounds, by I. N.
Barshchevskiy.

RUSSIAN, per, Zhur Fiz Khim, Vol. XXXIII, No 5,
1959, pp 1071-1079.

DSIR LLJ RTS 1537

Sci - Chem

Sep 60

125,547

Investigation of the Rate of Burning of American
Fuel in a Turbulent Flow. ID 2132569

RUSSIAN, FOR, SHIP THE China, Vol. XXVIII, 1959.
pp 1080-1086

Sci - Phys

Oct 59

ACER, H-100
ATS-05-37R
ATS/RD-2472

110,347

Vasil'yev, S. S.
KINETIC ANALYSIS OF CHAIN REACTIONS. V.
REARRANGEMENT OF THE FORM OF THE SOLU-
TIONS OF THE BASIC EQUATIONS OF THE KINET-
ICS OF CHAIN REACTIONS FOR THE CASE OF
TWO ACTIVE SPECIES (Kineticheskiy Analiz
Tsepnykh Reaktsii. V. Preobrazovaniye Formy
Reshenii Osnovnykh Uravnenii Kinetiki Tsepnykh
Reaktsii Silya Sluchaya Dvukh Aktivnykh Chastits).
Jan 61 [16]p. 9 refs. RTS 1739.
Order from LC or SLA ml\$2.40, ph\$3.30 61-15732

Trans. of Zhur[na]l Fiz[icheskoy] Khim[ii] (USSR)
1959, v. 33, no. 5, p. 1100-1110.

Transformations of the form of the solutions of the
basic differential kinetic equations of chain reactions
for the case of two active species led to expressions
that directly show how the kinetic development of
chain reactions is affected by the magnitudes of the
generalized rate constants of the initial kinetic
(Chemistry--Physical, TT, v. 5, no. 12) (over)

61-15732

- I. Chemical reactions--
Mathematical analysis
2. Title: Chain reactions
I. Vasil'yev, S. S.
- II. RTS-1739
- III. Department of Scientific and
Industrial Research
(Gt. Brit.)
- IV. Title: Rearrangement ...

10064

Office of Technical Services

Heat of Adsorption in the Electronic Chemical
Sorption Theory, by Sh. M. Kogan, V. B.
Sandomirskiy, 8 pp.

RUSSIAN, Dokl. Akad. Nauk SSSR, Vol. 128, 1959,
pp 1129-1133.

AEC Tr-4299

Sci - Phys, Chem
21 Feb 61

140,948

Fedosheva, O. P. and Tunitskii, N. N.
STUDY OF THE KINETICS OF ION EXCHANGE
SORPTION. III. COMPLETE EXCHANGE BETWEEN
MONOVALENT AND DIVALENT ION IN RESINS.
[1961] [9]p. 4 refs.

Order from OTS or SLA \$1.10

62-10307

Trans. of Zhurnal Fizicheskoi Khimii (USSR) 1959,
v. 33, [no. 5] p. 1140-1146.

DESCRIPTORS: *Ion exchange resins, Resins, Hydro-
gen, Sodium, Calcium, Strontium, Ions, Exchange
reactions, Reaction kinetics, Diffusion, Adsorption.

The effect was studied of the cationic form of the resin
on the diffusion coefficients of ionic microcomponents.
It is shown that monovalent as well as divalent ions
diffuse faster in a resin containing monovalent ions
than in a resin containing divalent ions. Swelling of
(Chemistry-Physical, TT, v. 7, no. 10) (over)

62-10307

I. Fedosheva, O. P.
II. Tunitskii, N. N.
III. Title: Complete ...

Office of Technical Services

19

(DC-3592).

Boris Vasil'yevich Yerofeyev (On his 50th
Birthday), by S. V. Markevich, 4 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXIII, No 5,
1959, pp 1155, 1156.

JPRS 3368

USSR

Biog

Jun 60

117,571

Study of the Oxidation of Cyclohexane by
the Method of Inhibition During the Reaction,
by E. T. Denisov, 13pp.

RUSSIAN, per, Zhurnal Fizicheskoi Khimii,
Vol 33, 1959, No 6, pp 1198-1208.

CFSTI TT-64-18412

323,208

Sci - Chemistry
Apr 67

Presentation of the Equation of State of Real
Gas in Explicit Form in Terms of the Independent
Variables T and v , by V. A. Zagoruchenko,
P. M. Kessel'man, 8 pp.
RUSSIAN, per, Zhurnal Fizicheskoy Khimiy, Vol 33,
No 6, 1959, pp 1221-1229.
NTIS TT 69-55051

Sci/Chem
Dec 70

Investigation of the Surface State of Nickel
and Iron Cathodes From the Drop in Potential
After Disconnecting the Current, by

V. E. Past, Z. A. Iofa.

RUSSIAN, per, Zhurnal Fizicheskoy Khimii,
Vol 33, No 6, 1959, pp 1230-1237.

NASA TT F-11,770

Sci
Dec 63

371,123

A Study of the Wetting of Solid Surfaces of Refractory Oxides by Liquid Metals, by V. N. Yerenenko, Yu. V. Naytlich.

RUSSIAN, per, Zhurnal Fiz Khimii, Vol 33, No 6, 1959, pp 1233-1245.

CIA X-5779

Oct 68

369,408

Investigation of the Influence of the Structure of Unsaturated Compounds on Their Reactivity in Copolymerization Processes. Part 2. The Effect of Different Substituents Attached to the Nitrogen on the Reactivity of Methacrylamide Derivatives with Respect to a Certain Type of Radical, by G. A. Petrova, G. A. Shtraykman, 9 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXIII, No 6, 1959, pp 1246-1252.

195, 969

AT3-2344R
AT3/RJ-2450

Sci
Vol 4, No 10, May 62

The Effect of Ionizing Radiation of the Corrosion of
Metals Under Atmospheric Conditions, 11 pp.

110 390 3902
RUSSIAN, per, Zhur Fiz Khim, Vol XXXIII, No 6,
1959, pp 1256-1262.

ACST, H-4619

Sci - Chem

Nov 59

TIC/IE-4 T-2171

100,509

61-15219

Zosimovich, D. P. and Bogatova, N. F.
ELECTRO DEPOSITION OF ZINC IN THE PRE-
SENCE OF SMALL QUANTITIES OF ANTIMONY AND
COBALT (Elektroliticheskoye Vydeleye Tsinka v
Prisutstvi Mal'kh Kolichestv Sur'my i Kobal'ta).
Nov 60 [8]p. 10 refs. RTS 1692.
Order from LC or SL: m51.80, p51.80 61-15219

Trans. of Zhurnal Fizicheskoy Khimii (USSR) 1959,
v. 33, no. 6, p. 1324-1327.

It is established that in the presence of small quanti-
ties of antimony (0.05 mg/l) and cobalt (20 mg/l)
during the electrolytic deposition of zinc an increase
in polarisation is observed. When cobalt is present in
solutions containing 0.2 mg/l of antimony, the process
is determined by the presence of antimony. Antimony,
introduced into the electrolyte either as potassium
antimonyl tartrate (tartar emetic) or as antimony
sulphate, has the same influence on the course of the
polarisation curves. It is suggested that small
(Chemistry--Physical, TT, v. 5, no. 7) (over)

- I. Zinc--Electrodeposition
- I. Zosimovich, D. P.
- II. Bogatova, N. F.
- III. RTS-1692
- IV. Department of Scientific
and Industrial Research
(Gt. Brit.)

148,730

Office of Technical Services

Phase Equilibria in the System Melamine-Ammonia,
by I. P. Krichevskiy, G. D. Efremova,

RUSSIAN, per, Zhur Fiz Khim, Vol XXXIII, No 6,
1959, pp 1328-1335.

ATE RJ-2040

Sci - Chem
Dec 59

103,974

A Method for Calculating Vapor-Liquid Equilibria
of Tricomponent Systems, by V. B. Kogan, V. M.
Safronov, 11 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXIII, No 6,
1959, pp 1353-1359.

SLA 60-17729

Sci

129,401

OTS, Vol IV, No 3

Sep 60

61-13488

Shilov, A. Ye. and Sabirova, R. D.
MECHANISM OF THE PRIMARY ACT OF DECOM-
POSITION OF CHLORINE DERIVATIVES OF METH-
ANE. I. DECOMPOSITION OF CARBON TETRACHLO-
RIDE AND METHYL CHLORIDE (Mekhanizm Pervich-
nogo Akta Termicheskogo Raspada Khlorproizvodnykh
Metana). Nov 60 [17]p. 19 refs. RTS 1527.
Order from LC or SLA ml\$2.40, ph\$3.30 61-13488

Trans. of Zhurnal Fizicheskoy Khimii (USSR) 1959
[v. 33] no. 6, p. 1365-1373.

The kinetics of the thermal decomposition of carbon
tetrachloride and methyl chloride in jets was investi-
gated. The primary decomposition in both cases taken
place with the disruption of a chlorine atom. A mech-
anism for the secondary reaction is suggested, which
agrees with the products of reaction and the kinetic
data. (Author)

(Chemistry--Physical, TT, v. 5, no. 4)

- I. Chlorides--Decomposition
- I. Shilov, A. Ye.
- II. Sabirova, R. D.
- III. Title: Decomposition...
- IV. RTS-1527
- V. Department of Scientific
and Industrial Research
(Gt. Brit.)

143,240

Office of Technical Services

63-12262

Kazanski, K. S., Entelis, S. G., and Chirkov, N. M.
THE SOLUBILITY OF GASEOUS ISOBUTYLENE IN
WATER. [1962] 6p.

Order from ATS \$8.95

ATS-97P64R

Trans. of Zhurnal Fizicheskoi Khimii (USSR) 1959,
v. 33, no. 6, p. 1409-1413.

DESCRIPTORS: *Butenes, Gases, Solubility, Water

- I. Kazanski, K. S.
- II. Entelis, S. G.
- III. Chirkov, N. M.
- IV. ATS-97P64R
- V. Associated Technical
Services, Inc., East
Orange, N. J.

ATL RJ-3942

ALL 872.2. 1964 (1110)

Handwritten: *Trill R. Co. Inc.*
Orange, N.J. 07070

(Chemistry--Physical, TT, v. 9, no. 2)

Office of Technical Services

Absorption Properties of Cellulose in the Process of
Its Thermal Decomposition, by T. V. Gatovskaya.

RUSSIAN, per, Zhur Fiz Khim., Vol 33, No 6, 1959,
pp 1418-1421.

*NTIS TT 71-51001

July 71

Marinin, V.A.
VARIATION THERMIQUE DU COEFFICIENT DE DIFFUSION DE CERTAINES SUBSTANCES DANS LE TETRACHLORÉTHANE ET LE TETRABROMÉTHANE (Temperature Relation of the Diffusion Coefficient of Substances in Tetrachloro- and Tetrabromoethane). 5p. 5refs. CNRS-VIII bis 690
Order from OIS, ETO or CNRS \$0.80 TT-62-26386

Trans. in French of Zhurnal Fizicheskoi Khimii (USSR) 1959, 33, no. 6, p. 1430-1433.

DESCRIPTORS: Bromine compounds, Chlorine compounds, Hydrogen compounds, Carbon compounds, *Diffusion, Heat, *Ethanes.

(Chemistry--Physical, TT, v. 11, no. 7)

DT

TT-62-26386

I. Marinin, V.A.
II. CNRS-VIII bis 690
III. Centre National de la Recherche Scientifique, Paris

Office of Technical Services
European Translations Centre

The Thermodynamics of the Superstate, by V. K.
Semanchenko, 8 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXIII, 1959,
pp 1440-1445.

ABC Tr-4271

Sci - Phys
9 Mar 61

141,572

Applying Thermodynamics of Irreversible Processes
to the Phase Theory. The Case of the Two Component
Two-Phase System at Constant Temperature and
Pressure, by Kiril A. Popov.
RUSSIAN, per, Zh. Fiz Khim, Vol 32, 1958,
pp 1638-47.
WAPD-tr-57

Sci - Chem
Oct 67

342,701

Ion Hydration and Its Mechanism. The Structure
and Dimensions of Hydrated Ions, by A. F.
Kapustinskiy, 15 pp.

RUSSIAN, per, Zhur Fiz Khim, 1953, Vol XXXII,
No 7, pp 1648-1657.

SLA 60-17725

Sci

3 Mar 62

186, 171

61-23816

Andreeva, V. V. and Shishakov, N. A.
DE L'ÉPAISSEUR DES PELLICULES D'OXYDES SUR
CERTAINES MÉTAUX SELON LES DONNÉES DES
ÉTUDES ELECTRONOGRAPHIQUES ET OPTIQUES
[On the Thickness of Oxide Films on Some Metals from
the Data of Electron Diffraction and Optical Investiga-
tions] tr. by de Trezvinaky. 22 Mar 60 [8]p. 5 refs.
CEA Trans. no. R 855 (text in French).
Order from OTS or SLA \$1.10

61-23816

Trans. in French of Zhurnal Fizicheskoi Khimii
(USSR) 1958, v. 32, no. 7, p. 1671-1672.

A translation in English is available from LC or SLA
at \$1.80, plus \$1.80 as ABC-tr-3955, Jan 60, 3p.

DESCRIPTORS: *Metals, *Oxidation, *Thin films,
*Oxides, Electron diffraction analysis, Optical analysis
Thickness.

For abstract see Technical Translations 3: 776, 1960.
(Chemistry-Physical, TT. v. 6, no. 6)

- I. Andreeva, V. V.
- II. Shishakov, N. A.
- III. CEA-tr-R855
- IV. Commissariat à l'Énergie
Atomique (France)

176696

Office of Technical Services

P. cat. my Arsenal
Req. 154-71

R--13363-D
28 June 1971

Zhur. Fiz. Khim. Vol. 32, pp. 1690-1691. (1958).
(Russian Journal of Physical Chemistry)

Medvedev, V. A.

Language: Russian

Please translate & type 1 copy only.

Method of Calculating the Thermodynamic Properties
of Some Compounds Without Knowledge of Their
Vibration Spectra, by Yu. P. Maslov, P. G.
Maslov. UICL

RUSSIAN, per, Zhur Fiz Khim, Vol XXIII, 1958,
pp 1725-1725.

DSIR LLU M.1285

(loan)

111,137

Sol - Phys; Chem

Mar 60

Andreev, K.K., Glazkova, A.P. and others.
DECOMPOSITION THERMIQUE DES NITRO-ETHERS.
I. ETUDE DE LA CINETIQUE DE LA DECOMPOSITION
DE LA NITROGLYCERINE ET DE NITROGLYCOL A
L'AIDE DE LA METHODE MANOMETRIQUE (The
Thermal Decomposition of Nitro Esters. I. The Mano-
metric Method of Studying Nitroglycerin and Nitro-
glycol Decomposition Kinetics). 20p. 12 refs
CNRS-VIII bis 495.

Order from DTIC or CNRS \$1.60 TT-62-26532

Trans. in French of Zhurnal Fizicheskoi Khimii
(USSR) 1958, v.32, no.1, p.1116-1127.

DESCRIPTORS: Nitroglycerin; Decomposition;
Kinetic theory; Liquids; Vapors; Gases; Mano-
meters; Reaction kinetics

(Chemistry--Physical) TT, v. 11, no. 8)

TT-62-26532

I. Andreev, K.K.
II. Glazkova, A.P.
III. CNRS VIII bis 495
IV. Centre National de la Recherche
Scientifique, Paris

Office of Technical Services
European Translations Centre

The Approximation Computation of the Temperature
Function of Heat Capacity, by M. Kh. Karapet'-
yants, 14 pp.

RUSSIAN, per. Zhur Fiz Khim, Vol. XXXII, No 8,
1958, pp 1763-1773.

OTS 60-17719

Sci

30 Mar 62

188, 910

Kinetics of Catalytic Oxidation of Methylene,
by G. M. Panchenko, A. M. Matanokaya, 6 pp.

RESEARCH, par, Thur Pie Main (1958), Vol. 2222,
No 2, 1958, pp 1779-1783.

Assoc Tech Serv
571222

ATS/RJ-1535

Sci. Chem

May 59

Q13, V 1, No 6

86,347

The Mechanism of the Cathode Process in the
Electrolysis of a Fused $\text{NaCl}-\text{CaCl}_2$ Mixture on
an Iron Cathode, by L. Suskii, 22² 19 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII, 1958,
pp 1785-1795.

AEC Tr 3658

Sci - Phys

Jan 59

*Dept of Chemistry
T-17, E-17, W-303*

88,899

Catalytic Properties of Cerium Dioxide in the
Dehydrogenation and Dehydration of Alcohols
and in the Dehydrogenation of Cyclohexane, by
A. A. Melstopyatova, A. A. Balandin.

RUSSIAN, per, Zhur Fiz Khim, Vol. XXXII, No 8,
1958, pp 1831-1841.

ATS RJ-1903

Sci - Chem
Dec 59

521 60-10582

104,080

TT-62-28263

Antonovskaya, E.I. and Sukhotin, A.M.
 SUR LES PARTICULARITES DE L'ACTIVATION
 ANODIQUE DU CHROME ET DES ACHERS AU CHROME.
 I (Ob Osobennostyakh Anodnogo Aktivirovaniya Khroma
 i Khromistykh Stalok). II (Peculiarities of the Anodic
 Activation of Chromium and Chromium Steels). D 14p.
 (foreign text included) Paris, CNRS-VIII bis 584.
 Order from OIS, ETC of CNRS \$0.80 TT-62-28263

I. Antonovskaya, E.I.
 II. Sukhotin, A.M.
 III. CNRS-VIII bis 584
 IV. Centre National de la Recherche
 Scientifique, Paris

Trans. in English of Zhurnal Fizicheskoi Khimii (USSR)
 1958, v. 32, no. 5, p. 1742-1749.

Trans. in English available from AFS \$8.00 as
 AFS-63ME-11.

DESCRIPTORS: Chromium. Chromium alloys.
 Iron alloys. Steel. Anodic. Activation coefficient, pH.

(Metallurgy--Ferrous Metals, TT, v. 11, no. 8)

Office of Technical Services
 European Translation Centre

S-322/60

(NY-3919)

Studies of Thermodynamic Properties of Substances
With the Aid of the Method of Explosion in a
Spherical Bomb, by V. A. Medvedev, 12 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII, No 8, 1958,
pp 1851-1859.

JPRS 2842

Sci - Chem

Jul 60

120,061

Enthalpy of Corundum in the Temperature Range From
100 to 900° C, by K. Z. Gonselskiy, 8 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXIII, No 8, 1958,
pp 1859-1862.

Dept of Commerce
Natl Bur of Standards
Dr Henry Birnbaum

REKHX

Sci - Chemistry

Jun 59

89,476

Study of Initiation and Propagation of Spherical
Detonation, by W. B. Lyons.

RESEARCH, ENG, PHYS RES BIRM, Vol. XXXI, No 8,
1958, pp 1858-1873. 9073-191

May 2012/7-58

Sci - Physics
Feb 59

81,165

Leon'tev, E. A. and Luk'yanovich, V. M.
 TECHNIQUE DE PREPARATION DES CORPS POREUX
 ET PULVERULENTS POUR L'ETUDE AU MICRO-
 SCOPE ELECTRONIQUE PAR LA METHODE DES
 REPLIQUES [Tekhnika Preparirovaniya pri Provedenii
 Elektronno-Mikroskopicheskikh Issledovaniy Poristykh
 i Poroshkoobraznykh tel Metodom Replik] (Technique
 for Preparation of Porous Bodies and Powders for
 Electron Microscopic Study by the Replica Method) tr.
 by Bouchicot. 21 Jan 60 [14]p. 8 refs. CEA Trans.
 no. R 812 (text in French).

Order from OTS or SLA \$1.60

61-23784

Trans. in French of Zhurnal Fizicheskoi Khim[ia]
 (USSR)1958, v. 32, no. 8, p. 1922-1925

DESCRIPTORS: *Porous materials, *Powders, Electron

(Physics--Optics. TT, v. 6, no. 7)

61-23784

- I. Title: Replicas
- I. Leon'tev, E. A.
- II. Luk'yanovich, V. M.
- III. CEA-tr-R812
- IV. Commissariat à l'Energie Atomique (France)

Office of Technical Services

TT-62-28257

Fedorov, N.A.
PROPRIETES DES ELECTRODES DE VERRE AU
LITHIUM (Elektrode Svoistva Litsevyogo Stekla) (The
Electrode Properties of Lithium Glass). 19p. (foreign
text included) Refs. CNRS-VIII bis 487.
Order from OTS, EITC or CNRS \$0.80 TT-62-28257

Trans. in French of Zhurnal Fizicheskoi Khimii (USSR)
1958, v. 32, no. 9, p. 1951-1957.

DESCRIPTORS: *Glass electrodes, *Lithium,
Chemical properties, High temperature research,
Measurement.

(Chemistry-Physics I, TT, v. 11, no. 8)

- I. Fedorov, N.A.
- II. CNRS-VIII bis 487
- III. Centre National de la Recherche
Scientifique, Paris

Office of Technical Services
European Translations Centre

Investigation of Hydrogen Over-Voltage on
Germanium, by Ye. A. Yefimov, I. G. Yerusalinchik.
9 pp.

RUSSIAN, per, Zhur Fiz Khim USSR, Vol XXXII,
No 9, 1958, pp 1967-1970.

SLA 59-15004

Sci
Dec 59
Vol 2, No 4

103085

Bogoyavlenskii, P. S.
THE EFFECT OF LARGE MONOVALENT IONS ON
THE STRUCTURE OF WATER AND ITS INFLUENCE
ON THE SOLUBILITY OF SALTS. [1962] 8p.
Order from ATS \$13.50 ATS-00P65R

Trans. of Zhurnal Fizicheskoi Khimii (USSR) 1958,
v. 32, no. 9, p. 2035-2041.

DESCRIPTORS: *Water. Molecular structure, Ions.
*Salts, Solubility.

(Chemistry--Physical, TT, v. 9, no. 5)

63-12474

I. Bogoyavlenskii, P. S.
II. ATS-00P65R
III. Associated Technical
Services, Inc.,
East Orange, N. J.

ATE RJ-2742

Office of Technical Services

On the Passivity of Iron in Hot Concentrated
Alkaline Solutions, by A. M. Sukhotin, K. M.
Kertashova; UNCLASSIFIED

(7 pp.)

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII, No 9, 1958,
pp 2068-2072. 9876777

Navy 2365/NERL 757

Sci - Physics
Nov 59

CIA/PID X-3756

101,413

Phase and Volume Correlations in Liquid-Gas ~~SYSTEMS~~
Systems Under High Pressures. VI.. System
Cyclohexane-Hydrogen, by I. P. Krichevskiy,
P. T. Sorina, 11 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII,
No 9, 1958, pp 2080-2086.

SLA 60-14057

Sci
Vol IV, No 5
Mar62

188, 446

Verigin, N. N.
DIFFUSION A LA SURFACE D'UN CORPS SOLIDE
PLONGE DANS UN LIQUIDE [Diffuziya u Poverkhnosti
Tverdogo Tela, Nakhodyashchegosya v Zhidkosti]
[Diffusion at the Surface of a Solid Body Immersed in a
Liquid] tr. by M. Grekow. 22 June 60 [22]p. 10 refs.
CEA Trans. no. 3 984 (text in French).
Order from OTS or SLA \$2.60 61-15803

Trans. in French of Zhurnal Fizicheskoy Khimii
(USSR) 1958, v. 32, no. 9, p. 2097-2106.

DESCRIPTORS: *Diffusion, Solids, *Corrosion,
Concrete, Rock, Liquids.

(Chemistry--Physical, TT, v. 6, no. 2)

61-15803

I. Verigin, N. N.
II. CEA-tr-R984
III. Commissariat à l'Énergie
Atomique (France)

10870

Office of Technical Services

Investigation of the Adsorption of Ions on Platinised IR
Platinum by Measuring the Adsorption Potentials, by
A. D. Cornicheva, 13 pp.

RUSSIAN, per. Zhur. Fiz. Khim., Vol XXXII, No 9, 1956,
pp 2155-2160.

NDP-0-110

Sci
Nov 59

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Effect of Ionizing Radiation on Aqueous
Solutions of Potassium Nitrate, by
A. N. Kabalchi, V. A. Gramolin, 11 pp.
RUSSIAN, per, Zhurnal fizicheskoiy khimiy,
Vol 32, 1958, No. 9, pp 2149-2154.
AEC-RFP-TRANS P911263468-V

Sci/Chem
Oct 68

369,446

Dissolution Anodique des Metaux Dans les Sels Fondus,
II. Beryllium. (Anodic Dissolution of Metals in
Fused Salts. II. Beryllium), by M. V. Smirnov,
N. Ya. Tschukreyev, 21 pp.

RUSSIAN to FRENCH, per, Zhur Fiz Khim, Vol XXXII,
1958, pp 2165-2173.

Reverse Trans
CEA tr R-653

Nov 59

Cathodic Processes in the Deposition of
Beryllium From Fused Electrolytes, by M. V.
Smirnov, L. V. Ivanovskiy, 19 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII, 1958,
pp 2174-2182.

AEC Tr-4416

Sci - Phys
May 61

150,330

Phenomenes Accompagnant le Depot Electrolytique du
Beryllium a Partir des Sels Fondus. (Phenomena
Accompanying the Electrolytic Deposition of
Beryllium From Fused Salts), by M. V. Smirnov,
L. E. Ivanovskiy, 18 pp.

RUSSIAN to FRENCH, per, Zhur Fiz Khim, Vol XXXII,
1958, pp 2174-2182.

Reverse Trans
CEA tr R-661

Nov 59

Determination of the Specific Surface Areas of
Porous Bodies and of Powders by the Method of
Gaseous Flow Under Low Pressure, by S. G. Entelis,

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII, No 9,
1958, pp 2187-2191.

ATS RJ-2167

Sci - Chem

May 60

115,445

On the Nature of Cold Flame Phenomena, by H. S.
Eriksson, et al.

EN
RUSSIAN, per, Zhur Fiz Khim, Vol XLVII, No 9,
1978, pp 2224-2226.

Navy 1989/T-59

Sol. - Physics
Jan 79

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Polymerization of Chlorinated Styrene Derivatives and Their Copolymerization with Vinyl Chloride, by G. V. Tkachenko, L. V. Stupen', et al., 10 pp.

RUSSIAN, per, Zhur Fiz Khim, 1958, Vol XXXII, No 10, pp 2250-2255.

SLA 59-10445

Sci - Chem.
OTS I, 12
Jul 59

92, 262

Balashova, N. A.
ADSORPTION DES IONS BROME ET IODE SUR LE
PLATINE [Adsorbtziya ionov Broma i Yoda na Platina]
[Adsorption of Bromine and Iodine Atoms on Platinum] tr.
by Mme. Simon. 5 July 60 [21]p. 28 refs. CEA Trans.
no. R 964 (text in French).
Order from OTS or SLA \$2.60

61-23780

Trans. in French of Zhurnal Fizicheskoi Khimii
(USSR) 1958, v. 32, no. 10, p. 2266-2273.
A translation in English is available from MDP \$6.00 as
MDP B-151 [1959] 12p.

DESCRIPTORS: *Platinum electrodes, *Adsorption,
Bromine, Iodine, Ions.

For abstract see Technical Translations 3: 566, 1960.

(Chemistry-Physical, TT, v. 6, no. 6)

61-23780

I. Balashova, N. A.
II. CEA-tr-R964
III. Commissariat à l'Énergie
Atomique (France)

176685

Office of Technical Services

The Cause of the Anticorrosion Effect of Sodium
Nitrite on Iron, by N. A. Shishakov, 6 pp.

UNCLASSIFIED

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII, No 10,
1958, pp 2282-2286. 9676948

Navy 2366/MRL 758

Sci - Physics
Nov 59

CIA/PDD X-3757

101,415

Separation of the Photochemical From the Light-
Filtering Effect and Latent Photosensitization, by
Yu. S. Zuev.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII, No 10, 1958,
pp 2308-2314.

AFS RJ-1724

SLA 62-18171

Sci - *chem*
Jul 59

90,458

The Conversion of Methane to Acetylene in the
High Voltage Arc, Part I, by E. N. Yeremin, et al.

RUSSIAN, per, Zhur Fiz Khim, Vol XIXII, No 10,
1958, p 2315

Sci~~u~~ Chem
Jul 59

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Navy 2232/T-147

ATS-PT-1977

Applied Chem Lab

TG 2232/T-147

SMR 60-21928

92, 812

The Promoting Effect of Cobalt Iron on the Chromate
Ion Catalysed Decomposition of Hydrogen Peroxide, by
G. I. Petrova, G. A. Bogdanov.

RUSSIAN, per, Zhur Fiz Khim, Vol. XXVIII, No 10, 1958,
pp 2324-2332.

ILL NIS 2269 (*Long*)

Sci - M/M
Feb 64

248, 283

The Kinetics and Mechanism of Catalytic Conversions
of Acetylene, II, Some Questions of the Mechanism
of the Addition of Various Molecules to Acetylene, by
R. M. Frid, 15 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII, No 10,
1958, pp 2339-2346.

SIA 59-10444

ATS-RJ-1908

Sci - Chem

Sep 59

Vol 2, No 1

97,281

63-10216

Vagin, F. V. and Zhukhovitskii, A. A.
THEORY OF THERMAL-ADSORPTIVE SEPARATION.
26 June 61, 25p. 6 refs.
Order from OTS or SLA \$2.60 63-10216

Trans. of Zhurnal Fizicheskoi Khimii (USSR) 1958,
v. 32, no. 10, p. 2372-2373.

DESCRIPTORS: *Adsorption, Separation, Gases, Sorp-
tion, Chromatographic analysis, Thermal diffusion,
Equations of state, Theory.

A theory of thermal-adsorptive separation of gases
based on the assumption of pressure equality in the
sorbent layer is presented. In the case of an infinitely
great sorption rate and of absence of longitudinal dif-
fusion in the process of thermal desorption, a distribu-
tion of the adsorbed substance corresponding to the
temperature field is established, the motion velocities
(Chemistry--Physical, TT, v. 10, no. 3) (over)

I. Vagin, F. V.
II. Zhukhovitskii, A. A.

Office of Technical Services

On the Charging of a Semiconductor Surface^F
During Adsorption, by F. F. Vol'kenshteyn, 12 p.

RUSSIAN, per. Zhur Fiz Khim, 1958, Vol XXII,
No 10, pp 2383-2391.

DDF V-135

Sci
Mar 60
Vol 2, No 11

109,670

On the Nature of the Negative Temperature Coefficient of the Rate of Radical Combination, by A. D. Stepankhovich.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII, No 10, 1958, p 2415. - 5417.

Navy 2233/T-148
OTS 60-21787

Sci - Physics
Jul 59

92,813

The Radiation Oxidation of Nitrogen. III. Some Questions Concerning the Mechanism of the Reaction and a Comparison With Electrical Discharge Data, by M. T. Daitryev.

RUSSIAN, per, Zhur Fiz Khim, Vol XXII, No 10, 1958, p 2418. - 2423

Navy 2234/T-149

OTS 60-21784

Sci - Physics
Jul 59

92,815

The Present State of Research on the Phase Diagram
of the Boron-Carbon System, by G. V. Samsonov, 7 pp

RUSSIAN, per, Zhur Fiz Khim USSR, Vol XXXII, No 10,
1958, pp 2424-2429.

ATS-55L314

Sci
Dec 59
Vol 2, No 4

102, 778

Electrochemical Properties of Zirconyl
Phosphate. Part 1. Electrode Properties
of Membranes, by E. A. Materova.
RUSSIAN, per, Zhurnal Fizicheskoi Khimii,
Vol 38, 1964, pp 676-680.
NTC 72-10454-07D

Apr 72

The Rate of Desorption, by D. P. Timofeyev.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII, No 11,
1958, pp 2483-2486.

NEL M. 2577

Sci - Chem

173,878

Nov 61

61-20705

Tkachenko, G. V., Stupen, L. V. and others.
THE COPOLYMERIZATION OF VINYL CHLORIDE
WITH METHACRYLIC ACID ESTERS. [1961] [9]p.
10 refs.

Order from OTS or SLA \$1.10

61-20705

Trans. of Zhurnal Fizicheskoi Khimii (USSR) 1958,
v. 32, no. 11 [p. 2492-2499].

DESCRIPTORS: *Vinyl chlorides, Copolymerization,
Polymerization, *Acrylic acids, Acids, Esters,
Benzoates, Polymers, Molecular weight.

The copolymerization of vinyl chloride with methyl-,
n-butyl-, n-octyl methacrylates and vinyl benzoate
was studied. The rates of the copolymerization of
vinyl chloride with methacrylates and the molecular
weights of the polymers in all of the corresponding
monomers are less than with acrylates. These values
(Chemistry--Organic, TT, v. 7, no. 10) (over)

I. Tkachenko, G. V.
II. Stupen, L. V.

Office of Technical Services

Yeremin, Ye. N.
THE KINETICS OF ELECTROCRACKING OF METH-
ANE [Kinetika Elektrokrekninga Metana].
[1961] 7p
Order from ATS \$12.55

ATS-52N51R

Trans. of Zhurnal Fizicheskoy Khimii (USSR)
1958, v. 32, no. 11, p. 2543-2549.

DESCRIPTORS: *Methanes, Decomposition, Hydro-
carbons, Reaction kinetics.

(Chemistry--Physical, TT, v. 6, no. 3)

61-22612
(63-18012)

- I. Yeremin, Ye. N.
- II. ATS-52N51R
- III. Associated Technical
Services, Inc., East
Orange, N. J.

ATS-RJ-2892

170823

Office of Technical Services

Studies in the Mechano-Chemistry of Polymers
XXIII. Mechanical and Chemical Processes
in the Mastication of PVC, by A. A. Berlin,
et al.

RUSSIAN, per, Zhur Fiz. Khim, Vol XXXII,
No 11, 1958, pp 2565-2570.

CHIRO

207,417

Sci - Chem
Aug 62

The Use of the Electric Arc in Calorimetry,
by A. F. Vorob'yev, S. M. Skuratov, 8 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII, No 11,
1958, pp 2580-2585.

SLA 60-13082

Sci
Aug 61

OTS, Vol III, No 12

163,132

S-85/60

(NY-3554).

On the Computation of a Single Coefficient of the
Separation of Isotopes for Equilibrium Processes,
by Ye. M. Kuznetsova, A. V. Makarov, G. M.
Panchenkov, 3 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII, No 11,
1958, pp 2641-2643.

JPRS 2476

114,730

Sci -- Chem
Apr 60

1/19/59

"The 40 th Anniversary of the Physico-Chemical
Institute imeni L. Ya. Karpov" /unsigned article/,
Zhurnal Fizicheskoy Khimii, Vol 32, No 12,
Dec 58, pp 2657-2662.

SIR T-16.

NCH-202 507

Field 7D

Sinytsin, Z. A.

THE INHIBITING EFFECT AND THE STRUCTURE OF
INHIBITORS. 2. THE INHIBITED POLYMERIZATION
OF METHYLACRYLATE. 1965, 16p. TC-1221.
Order from TC: \$15.00 as TC-1221

Trans. of Zhurnal Fiz. Khimii (USSR) v32 n12
p2663-73, 1958.

I. Trans-Chem
Knoxville, Tenn.
II. Title: Inhibited. . .

The Electrochemical and Corrosion Behavior of
Steel and Nickel Electrodes in Sulphuric Acid
Solutions Subjected to the Action of Rays, by
N. Ya. Bure, Ya. M. Kolotyrkin, G. S. Tyurikov,
11 pp.

RUSSIAN, per. Zhur Fiz Khim, Vol. XXXII, 1956,
pp 2679-2685.

AEC NP Tr-445

Sci . Chem

125,848

Relation Between the Energy Yield and the Kinetics
of a Radiation-Chemical Reaction, by S. Ya.
Peshetskiy, M. T. Daitryev. UNCLASSIFIED

RUSSIAN, per, Zhur Fiz Khim, Vol XIXII, No 12,
1958, pp 2586-2589. 92,828

Navv 2211/BRL 727

Sci - Physics, Chem
Jul 59

92, 828

Shatenshtein, A. I. and Izrallevich, E. A.
COMPARISON OF DEUTERO EXCHANGE AND
METALATION OF ALKYL BENZENES. [1962] [9]p.
30 refs.

Order from OTS or SLA \$1.10

63-15923

Trans. of Zhurnal Fizicheskoi Khimii (USSR) 1958
[v. 32] no. 12, p. 2711-2716.

DESCRIPTORS: *Metalorganic compounds, *Potassium
compounds, Amides, *Alkyl radicals, *Benzenes,
*Deuterated compounds, Exchange reactions, *Metal-
tion, Reaction kinetics

The rate of deuterio exchange between a solution of
potassiumide (0.02N) in liquid deuterioammonia and the
aromatic rings and alkyl groups of the following alkyl
benzenes: $\text{C}_6\text{H}_5\text{CH}_3$, $\text{iso-C}_6\text{H}_5\text{C}_3\text{H}_7$, $\text{tert-C}_6\text{H}_5\text{C}_4\text{H}_9$,
and $n\text{-C}_6\text{H}_5\text{C}_4\text{H}_9$ has been measured at 25, 10, and 0°C.
It has been found that deuterio exchange and the metal-
(Chemistry--Physical, TT, v. 10, no. 1) (over)

63-15923

I. Shatenshtein, A. I.
E. Izrallevich, E. A.
III. Translations, New York

Office of Technical Services

The Use of Radioactive Sulfur to Study Hydro-
formylation of Acrylonitrile-Synthesis Catalyst, by
M. E. Bulatnikova, L. O. Apal'dayeva, M. I.
Tikhonov.

RUSSIAN, PER. Zhurnal Khim. Vol. 31, No. 12,
1958, pp 2717-2724.

OSIR 240 M.791 (Cm)

Dept of Interior
THT, B57, Ro 162

Sci - Chem
Jan 60

SLA 59-18543 102046

Dissociation of Deuteriooctane Under Elec-
tron Impact, by M. V. Goryunov, M. V.
Mikhomirov, 10 pp.

RUSSIAN, Zhur Fiz Khim, Vol XXXII,
No 12, 1956, pp 2731-2738.

ATS-08M42R

ATS/RS-2545

Ref
Vol IV, No 11
Jun 62

199,250

Boraskov, G. K.
CATALYST INTERACTION WITH THE REACTION
SYSTEM. [1963] 10p. 20 refs.
Order from OTS or SLA \$1.10

63-14330

Trans. of Zhurnal Fizicheskoi Khimii (USSR) 1958,
v. 32 [no. 12] p. 2739-2747.

DESCRIPTORS *Catalysts, Chemical reactions, Theory,
Metals, Oxides, Reaction kinetics.

In the course of a catalytic reaction, besides the
action of the catalyst on the reagents, there is always
an action on the reacting system on the catalyst
leading to changes in its composition and consequently
in its properties. To each composition of the reacting
mixture there corresponds a definite, steady state
composition of the catalyst depending upon the ratio of
the rates of interaction between the catalyst and the
(Chemistry--Physical, IT, v. 10, no. 3) (over)

63-14330

I. Boraskov, G. K.

Office of Technical Services

The Conversion of Methane to Acetylene in a High-
Voltage Arc. Part 2. The Effect of Hydrogen, by
E. H. Eremin,

RUSSIAN, Izv. Akad. Nauk SSSR, Vol. XXXII, No 12, 1958,
pp 2767-2771.

ATS RJ-2073

ATS-162378

Sci - Chem

May 60

115379

The Mechanism of Formation of H_2O and H_2O_2 in the
Reaction of Hydrogen Atoms With Oxygen Molecules,
by L. I. Abramenko, R. V. Kolesnikova. USSR

RUSSIAN, per, Zhur Fiz Khim, Vol XXII, No 12,
1958, pp 2780-2786.

DSIR LUJ RFB 121B

(17a. 6a.)

Sci - Chem
Nov 59

100,787

The Kinetics of Consecutive Reactions in a
Recycle-Flow System, by S. V. Dobrovol'skiy,
V. Ya. Polotnyuk.

RUSSIAN, par, Zhur Fiz Khim, vol XXXIII, No 12,
1958, pp 2732-2735.

SL 168-1503
ATS RJ-2037

Sci - Chem
Dec 59

103,771

Bochnar, D. A., Gambaryan, N. P., Stankevich, I. V.,
and Chistyakov, A. L.

A QUALITATIVE DETERMINATION OF THE STABILITY OF HETEROCYCLIC SYSTEMS WITHIN THE FRAMEWORK OF THE HUCKEL APPROXIMATION. [1963] 9p 2 refs

Order from OTS, SLA, or ETC \$1.10 TT-63-20684

Trans. of Zhur[asil] Fiz[ichesko] Khim[iti] (USSR) 1958, v. 32 [no. 12] p. 2797-2802. (Abstract available)

DESCRIPTORS: *Heterocyclic compounds, *Organoboranes, Chemical bonds, Stability, Atomic orbitals, Polynomials.

A purely qualitative method is proposed for deciding, within the framework of the Huckel approximation, the question of the presence of a "closed electron shell" in conjugated systems and, in this sense, the question of the relative stability of such systems. This method con- (Physics--Molecular, IT, v. 11, no. 3) (over)

TT-63-20684

1. Title: Borepin
2. Title: Huckel approximation
3. Title: Molecular orbital theory
- I. Bochnar, D. A.
- II. Gambaryan, N. P.
- III. Stankevich, I. V.
- IV. Chistyakov, A. L.

Office of Technical Services

Thermochemistry of Isotopes. 2 I. Calorimetric
Investigation of the Heat of Formation of Hydrogen
Sulfide and Deuterium Sulfide, by A. P. Kapustinskiy,
R. T. Kan'kovskiy, 12 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII, No 12, 1958,
pp 2810-2816.

US Dept of Commerce
National Bur of Standards

Sci - Phys
May 61

148,772.

Filippov, Yu. V. and Yemel'yanov, Yu. M.
ELECTRICAL THEORY OF OZONIZERS. III. THE
ELECTRIC CURRENT IN OZONIZERS (Elektriches-
kaya Teoriya Ozonizatorov. III. Elektricheskiy Tok v
Ozonizatorakh) tr. by E. Feigl. [1960] 114p. (foreign
text included) 5 refs. Trans. no. 4850.
Order from LC or SLA ml\$2.40, ph\$3.30 61-15173

Trans. of Zhurnal Fizicheskoy Khimii (USSR) 1958,
v. 32, no. 12, p. 2817-2823.

The previous theory (Zhur. Fiz. Khim. 31: 1628, 1957)
is used to derive expressions for the dependence of the
effective and mean values of the strength of the current
flowing through the ozonizer on (1) the parameters of
the apparatus and (2) the magnitudes of the voltage ap-
plied to the terminals. (See also 59-10012)

(Engineering--Electrical, TT, v. 5, no. 5)

61-15173

1. Electric discharges
2. Ozone--Production
3. Title: Ozonizer
- I. Philippov, Yu. V.
- II. Yemel'yanov, Yu. M.
- III. Title: Electric...
- IV. CSIRO Trans-4850
- V. Commonwealth Scientific
and Industrial Research
Organization (Australia)

146,964

Office of Technical Services

12 216

US-14 *UK-12

GUR'EV M. V., TIKHOMIROV M. V., TUNITSKII N. N.

The mass spectra of large molecules

*Zhur. Fiz. Khim., 32, No. 12, p. 2847 (1958)

US-14: ATS-09M42R. \$ 2.75 - English

*UK-12: English

E u r a t o m
ATS/RS-2546

The Synthesis of Some Deuterated Aromatic Hydrocarbons and their Raman Spectra, by A. I. Shatenshtein.

RUSSIAN, per, Zh Fiz Khim, Vol 32, No 1, 1958, pp 146-151.

HEW-NIH

12-4-67

Sci/B & M
Feb 68

349,425

ZHURNAL FIZICHESKOY KHIMII - monthly
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No 7, pp 1674-1675 JPRS 3368

No 12, pp 2755-2766 JPRS 2736

Viscosity Measurements with Microamounts of
Liquid, by V. S. Bondarenko.
RUSSIAN, per, Zhurnal Fizicheskoi Khimii, Vol 38,
1964, pp 1273-1275.
NTC 71-14363-07D

Mar 72

(NY-4844)

CO + O₃ Flame Radiation and the Mechanism of This
Reaction, by M. Zatsiorskiy, V. Kondrat'ev, S.
Solnyshkova, 14 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XIV, No 12, 1960,
pp 1521-1527.

JPRS 7532

Sci - Chem

Feb 61

139.356

Electrosynthesis of O₃. I. Kinetics of the
Synthesis of O₃ Under Flow Conditions, by
Yu. V. Filippov, Yu. H. Emel'yanov.
RUSSIAN, per. Zh. Fiz Khim, Vol 35, 1961,
pp 407-415.
AEC-911-Trans-522

Sci-Chem
Jan 68

348,751